

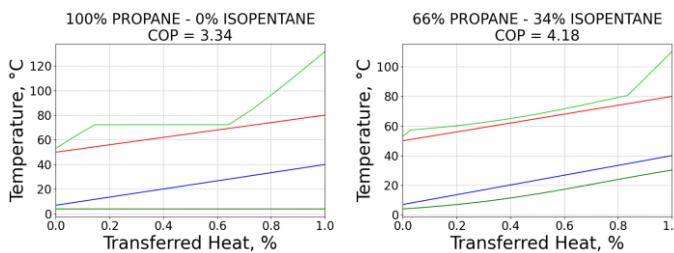
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## 1. Introduction

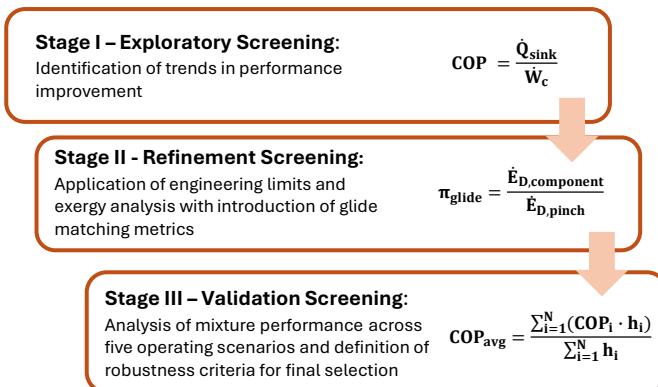
- Constant-temperature phase change of pure refrigerants often leads to poor temperature matching and exergy losses<sup>1</sup>.
- Zeotropic mixtures exhibit **temperature glide** during phase change caused by changing liquid and vapor compositions, enabling improved thermal matching and reduced irreversibilities.
- Oil circulation and fractionation can shift the circulating composition, introducing uncertainty around the nominal mixture ratio and affecting performance.



## 2. Research Objective

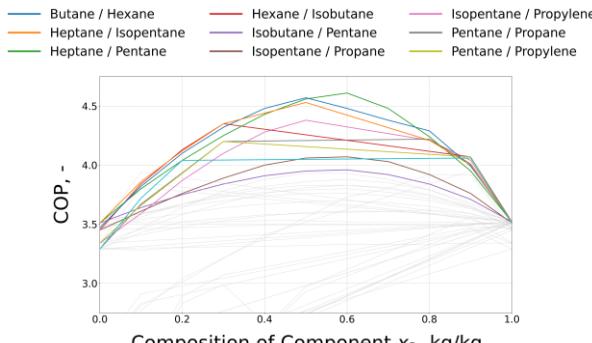
To develop a multi-stage screening framework that identifies zeotropic mixtures offering robust performance improvements across multiple operating scenarios, rather than optimizing mixtures for a single, case-specific application.

## 3. Screening Framework

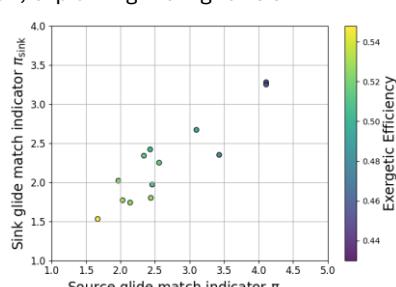


## 4. Performance improvements

- Hydrocarbon mixtures achieved **COP improvements of roughly 10–25%** compared to their pure-fluid counterparts, while maintaining acceptable pressures and discharge temperatures across all operating cases.



- Mixtures with good **glide match indicators**<sup>2</sup>, especially on the heat source side, significantly reduce avoidable exergy destruction, explaining the higher COP.

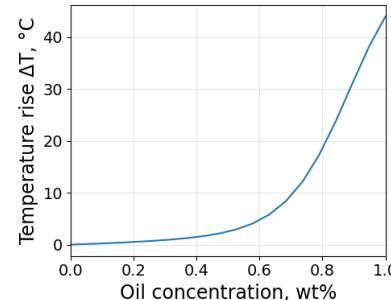


## 5. Oil Solubility Differences

- The presence of oil in the refrigerants mixture alters the phase equilibrium, which is described as a non-ideal multicomponent system:

$$\phi_{r,i} y_{r,i} P = \gamma_{r,i} x_{r,i} f_{r,i}^0$$

- At constant pressure, the saturation temperature in the presence of oil is higher than that of the oil-free single fluid refrigerant or refrigerant blend<sup>3</sup>.



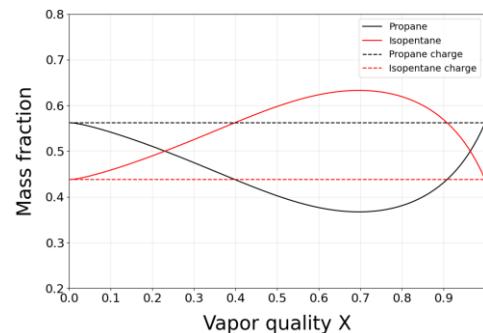
- Differences in oil solubility between mixture components can create absorption of one component into the oil which results in composition shift.

## 6. Slip Effects

- The composition during phase change may deviate from the charged composition as a consequence of **liquid hold-up**. The composition shift is expressed as<sup>4</sup>:

$$\delta = C_{j,i} - D_{j,i} = \frac{X_j(1 - X_j)(S_j - 1)(x_{j,i} - y_{j,i})}{(1 - X_j)S_j + X_j}$$

- As vapor quality increases, preferential evaporation leads to enrichment of the circulating fluid in the low-boiling component, causing a deviation from the nominal charged composition.



## 7. Conclusion

- Systematic screening of mixtures across a range of pressures and temperatures, is required to ensure robust performance improvements.
- The combined effects of oil solubility and liquid holdup introduce uncertainty in the circulating mixture composition, which cannot be captured by nominal charge specifications alone.

## 8. Acknowledgments

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## 9. References

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